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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/528,125	11/21/2005	Jorg Heller	STERN21.003APC	8936
20995 7590 03/29/2007 KNOBBE MARTENS OLSON & BEAR LLP				
2040 MAIN STREET			WONG, EDNA	
FOURTEENTH FLOOR IRVINE. CA 92614			ART UNIT	PAPER NUMBER
,		·	1753	
SHORTENED STATUTOR	Y PERIOD OF RESPONSE	NOTIFICATION DATE	DELIVERY MODE	
3 MO	NTHS	03/29/2007	ELECTRONIC	

### Please find below and/or attached an Office communication concerning this application or proceeding.

If NO period for reply is specified above, the maximum statutory period will apply and will expire 6 MONTHS from the mailing date of this communication.

Notice of this Office communication was sent electronically on the above-indicated "Notification Date" and has a shortened statutory period for reply of 3 MONTHS from 03/29/2007.

Notice of the Office communication was sent electronically on above-indicated "Notification Date" to the following e-mail address(es):

jcartee@kmob.com eOAPilot@kmob.com

		th	
	Application No.	Applicant(s)	
	10/528,125	HELLER ET AL.	
Office Action Summary	Examiner	Art Unit _	
	Edna Wong	1753	
The MAILING DATE of this communication appeariod for Reply	ppears on the cover sheet with th	e correspondence address	
A SHORTENED STATUTORY PERIOD FOR REP WHICHEVER IS LONGER, FROM THE MAILING  - Extensions of time may be available under the provisions of 37 CFR 1 after SIX (6) MONTHS from the mailing date of this communication.  - If NO period for reply is specified above, the maximum statutory perior  - Failure to reply within the set or extended period for reply will, by statu. Any reply received by the Office later than three months after the mail earned patent term adjustment. See 37 CFR 1.704(b).	DATE OF THIS COMMUNICAT 1.136(a). In no event, however, may a reply but d will apply and will expire SIX (6) MONTHS fute, cause the application to become ABANDO	ON. e timely filed rom the mailing date of this communication. DNED (35 U.S.C. § 133).	
Status			
1) Responsive to communication(s) filed on 27	February 2007.		
2a)⊠ This action is <b>FINAL</b> . 2b)☐ Th	is action is non-final.		
3) Since this application is in condition for allow	·		
closed in accordance with the practice under	Ex parte Quayle, 1935 C.D. 11	453 O.G. 213.	
Disposition of Claims			
4) ☐ Claim(s) 1-18 is/are pending in the applicatio 4a) Of the above claim(s) is/are withdr 5) ☐ Claim(s) is/are allowed. 6) ☐ Claim(s) 1-18 is/are rejected. 7) ☐ Claim(s) is/are objected to. 8) ☐ Claim(s) are subject to restriction and/	awn from consideration.		
Application Papers			
9)☐ The specification is objected to by the Examir	ner.		
10)☐ The drawing(s) filed on is/are: a)☐ ac	ccepted or b) objected to by the	e Examiner.	
Applicant may not request that any objection to the		· ·	
Replacement drawing sheet(s) including the corre		•	
Priority under 35 U.S.C. § 119			
a) Acknowledgment is made of a claim for foreign a) All b) Some * c) None of:  1. Certified copies of the priority documents.  2. Certified copies of the priority documents.  3. Copies of the certified copies of the priority application from the International Bure.  * See the attached detailed Office action for a list	nts have been received. nts have been received in Applic fority documents have been rece au (PCT Rule 17.2(a)).	cation No eived in this National Stage	
Attachment(s)  1) Notice of References Cited (PTO-892)	4)		
2) Notice of Draftsperson's Patent Drawing Review (PTO-948) 3) Information Disclosure Statement(s) (PTO/SB/08) Paper No(s)/Mail Date	5) Notice of Inform 6) Other:		

This is in response to the Amendment dated February 27, 2007. The text of those sections of Title 35, U.S. Code not included in this action can be found in a prior Office Action.

## Response to Arguments

### Specification

The abstract of the disclosure has been objected to because the word "Said" is used in line 2.

The objection of the abstract of the disclosure has been withdrawn in view of Applicants' amendment.

#### Claim Objections

Claims 1 and 5 have been objected to because of minor informalities.

The objection of claims 1 and 5 has been withdrawn in view of Applicants' amendment.

## Claim Rejections - 35 USC § 112

Claims **1-18** have been rejected under 35 U.S.C. 112, second paragraph, as being indefinite for failing to particularly point out and distinctly claim the subject matter which applicant regards as the invention.

The rejection of claims 1-18 under 35 U.S.C. 112, second paragraph, has been

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withdrawn in view of Applicants' amendment.

## Claim Rejections - 35 USC § 103

Claims 1-18 have been rejected under 35 U.S.C. 103(a) as being unpatentable over DE 198 55 666 ('666) and Lehmkuhl et al. (US Patent No. 6,652,730 B1) in combination with Dotzer et al. (US Patent No. 3,969,195).

The rejection of claims 1-18 under 35 U.S.C. 103(a) as being unpatentable over DE 198 55 666 ('666) and Lehmkuhl et al. in combination with Dotzer et al. is as applied in the Office Action dated November 27, 2006 and incorporated herein. The rejection has been maintained for the following reasons:

Applicants state that the combination of the cited references is not obvious, and is surprising, since it cannot be deduced from these documents that when the electrolytes according to Lehmkuhl are used, it is possible to expand the group of materials that can be treated in a single bath.

In response, Dotzer teaches that the coating and surface finishing of articles made of light metals, particularly beryllium, magnesium, aluminum, titanium and zinc and their alloys, is often necessary, because they are relatively base metals whose surfaces rapidly develop <u>a fundamentally oxidic coating</u> when exposed to the atmosphere. Such an oxidic coating usually protects the underlying metal against further corrosive attacks. However, the surfaces of articles made of such metals properly cannot be finished or coated in aqueous or <u>protic media</u>, due to the

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characteristics of the metals and oxide coatings (col. 1, lines 15-25). An aprotic organoaluminum electrolyte medium can be used to remove the oxidic coating (col. 6, lines 30-42) and electroplate aluminum onto the surface of the article (col. 7, lines 7-16).

Lehmkuhl teaches the electrolytic deposition of aluminum on electrically conductive materials using an aprotic organo-aluminum electrolyte medium (col. 1, lines 8-13).

Since beryllium and aluminum would have been obvious as the electrically conductive materials disclosed by Lehmkuhl, one having ordinary skill in the art would have been able to deduce when the electrolytes according to Lehmkuhl are used.

Applicants state that Lehmkuhl neither mentions anything about the pretreatment procedure nor does it specify what kind of solution is used for the pretreatment.

In response, the rejection is not overcome by pointing out that one reference does not contain a particular limitation when reliance for that teaching is on another reference. *In re Lyons* 150 USPQ 741 (CCPA 1966). Moreover, it is well settled that one cannot show nonobviousness by attacking the references individually where, as here, the rejection is based on a combination of references. *In re Keller* 208 USPQ 871 (CCPA 1981); *In re Young* 159 USPQ 725 (CCPA 1968).

Furthermore, claim 1, lines 3-4, recite "*immersing* the material in an electrolytic bath comprising electrolyte for pretreatment, wherein said material is connected as an

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anode therein". The pretreatment is the immersing of the material as an anode in an electrolytic bath comprising electrolyte. "Immersing" means to plunge or dip into a fluid. There is no electrolytic reaction during the immersing. Thus, the kind of solution that is used for the pretreatment would not have affected the method.

Applicants state that the electrolyte used for the pretreatment is different from the presently claimed invention because the '195 patent describes the use of halogen containing electrolytes for this purpose.

In response, the electrolyte used for the pretreatment would not have been different from the presently claimed invention when the polarity is reversed in the electrolytes disclosed by Lehmkuhl.

Furthermore, the claims as presently written are open to having halogens in the electrolyte bath.

Applicants state that the fused electrolyte described in the cited document is not suitable for the electrodeposition of aluminum, magnesium or aluminum/magnesium layers on other materials.

In response, claim 1, line 1, recites "electrolytic coating of <u>a material</u>". Lehmkuhl teaches the electrolytic deposition of aluminum on <u>electrically conductive materials</u> using an aprotic organo-aluminum electrolyte medium (col. 1, lines 8-13).

There is no reason why the electrically conductive materials disclosed by

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Lehmkuhl are not suitable materials as presently claimed or suitable as other materials.

Applicants state that the methods described in the cited references are not suitable for the metal coating of magnesium, aluminum/magnesium alloys or zinc substrates.

In response, claim 1, line 1, recites "electrolytic coating of a material". There is no claim basis for magnesium, aluminum/magnesium alloys or zinc substrates. It is well settled that unpatented claims are given the broadest, most reasonable interpretation and that limitations are not read into the claims without a proper claim basis therefor. In re Prater 415 F. 2d 1393, 162 USPQ 541 (CCPA 1969); In re Zeltz 893 F. 2d 319, 13 USPQ 1320.

#### Response to Amendment

### Claim Objections

Claims 1, 4-5 and 9 are objected to because of the following informalities:

#### Claim 1

line 3, the word -- an -- should be inserted after the word "comprising".

### Claim 4

line 2, "%,-" should be amended to -- %- --.

Claim 5

line 2, the word -- bath -- should be inserted after the word "electrolyte" because

the 3.3 mol toluene claimed is the aprotic solvent being used as the solvent for the

electrolyte. Toluene is not the electrolyte.

Claim 9

line 3, the word -- bath -- should be inserted after the word "electrolyte" because

the 3.3 mol toluene claimed is the aprotic solvent being used as the solvent for the

electrolyte. Toluene is not the electrolyte.

Appropriate correction is required.

THIS ACTION IS MADE FINAL. Applicant is reminded of the extension of time

policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE

MONTHS from the mailing date of this action. In the event a first reply is filed within

TWO MONTHS of the mailing date of this final action and the advisory action is not

mailed until after the end of the THREE-MONTH shortened statutory period, then the

shortened statutory period will expire on the date the advisory action is mailed, and any

extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of

the advisory action. In no event, however, will the statutory period for reply expire later

than SIX MONTHS from the mailing date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Edna Wong whose telephone number is (571) 272-1349. The examiner can normally be reached on Mon-Fri 7:30 am to 4:00 pm.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Nam Nguyen can be reached on (571) 272-1342. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

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Edna Wong Primary Examiner Art Unit 1753

EW March 25, 2007